



Methylglyoxal - a causative agent of detrimental glucose oxidase modification in manuka honey



Európska únia
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INTRODUCTION

Honey is considered to be an important therapeutic agent possessing significant antibacterial properties. Hydrogen peroxide (H₂O₂) is one of the major antibacterial components, and is produced by glucose oxidase (GOX)-mediated conversion of glucose under aerobic conditions in diluted honey (Fig.1).

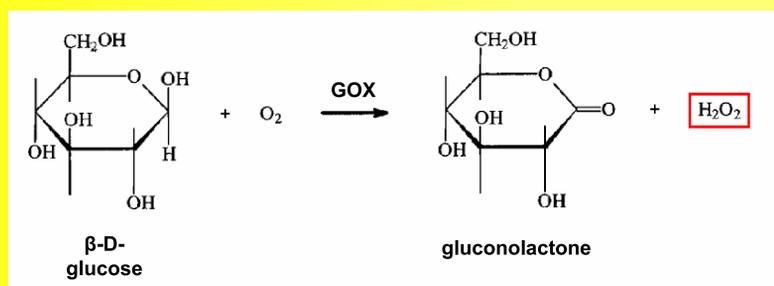


Fig.1 GOX-catalyzed production of H₂O₂ from glucose and oxygen. Taken and adapted from [1].

However, H₂O₂ does not accumulate in medical-grade manuka honey, and the pronounced antibacterial activity directly originates from methylglyoxal (MGO, C₃H₄O₂, Fig.2). Manuka honey is produced in New Zealand by domesticated European honeybees from the nectar of the manuka tree (*Leptospermum scoparium*). MGO originates from dihydroxyacetone (present in the nectar) upon storage of freshly produced honey.

AIMS

Recently, destructive effects of MGO on some proteins in manuka honey have been documented [2]. Based on this knowledge, we hypothesized that the inability to produce high levels of H₂O₂ could be associated with high reactivity of MGO with GOX enzyme (Fig.2). Accordingly, we investigated the effect of artificially added MGO on H₂O₂ accumulation in natural non-manuka honeys, which are capable to generate high levels of H₂O₂. Furthermore, we examined the effect of MGO on in vitro cross-linking of purified GOX and its biological activity.

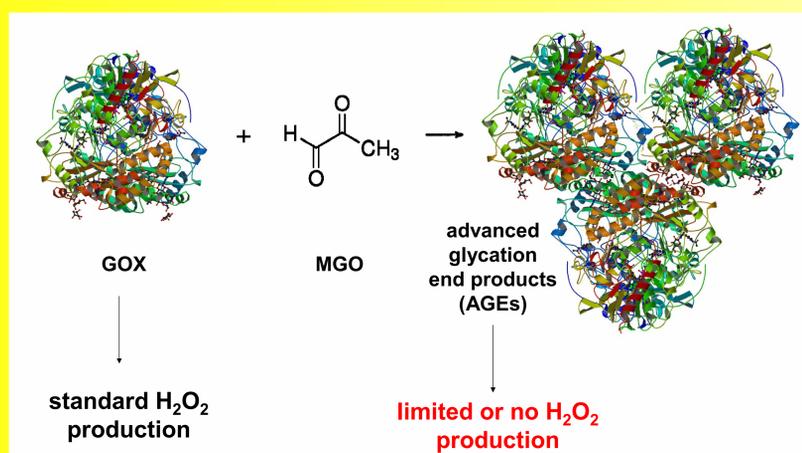


Fig.2 Proposed scenario of H₂O₂ production inhibition. Reaction between GOX and MGO presumably leads to the formation of advanced glycation end products (AGEs) with limited or completely lost ability for H₂O₂ production. 3D structure of GOX enzyme taken from [3].

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RESULTS

To evaluate the production of H₂O₂ we employed fluorometric resorufin assay (OxiSelect™ Hydrogen Peroxide/Peroxidase Assay Kit; Cell Biolabs). Among several tested honeys, the most potent producers were the two honeydew honeys (Table 1). Contrary to this, the MGO-treated honeys generated significantly lower amounts of H₂O₂ (Fig.3).

Table 1. H₂O₂ concentrations in different honey solutions (50% w/v) after 24 h incubation at 37°C.

Honey sample	Principal botanical origin	Sample age (year)	Geographic origin	H ₂ O ₂ concentration (μM)
honeydew 1	<i>Abies alba</i> Mill	2	Bardejov (Slovakia)	495.8 ± 9.1
honeydew 2	<i>Abies alba</i> Mill	2	Banská Štiavnica (Slovakia)	306.9 ± 6.8
acacia	<i>Robinia pseudoacacia</i>	1	Šahy (Slovakia)	36.3 ± 0.7
manuka	<i>Leptospermum scoparium</i>	3	New Zealand	78.9 ± 1.9
artificial	N/A	1	N/A	16.1 ± 1.2

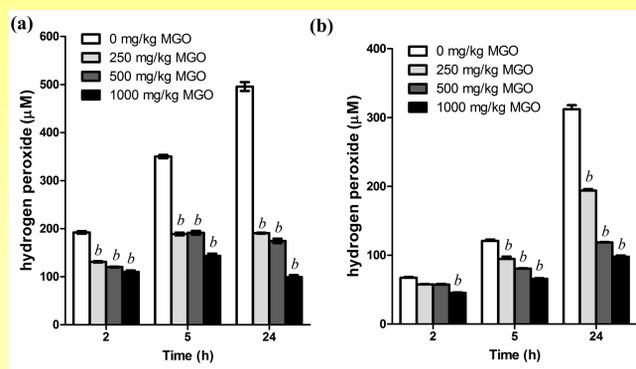


Fig.3 H₂O₂ concentrations in honeydew honey solutions (50% w/v) treated with MGO at different concentrations after 24 h incubation at 37°C. (a) honeydew honey (Bardejov) (b) honeydew honey (Banská Štiavnica) *b* P<0.001 vs. control (0 mg/kg MGO).

Furthermore, we observed formation of high molecular weight adducts on SDS-polyacrylamide (SDS-PAGE) gels when GOX was treated with MGO (Fig.4). This observation was confirmed by ultra high-performance liquid chromatography (UHPLC) analysis (Fig.5).

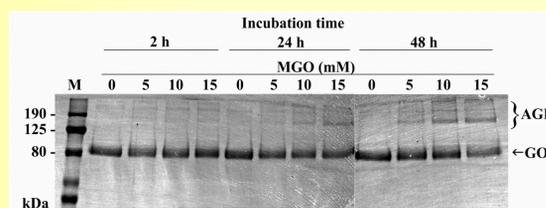


Fig.4 In vitro cross-linking of GOX by MGO - SDS-PAGE analysis. GOX at concentration of 1 mg/ml was incubated with or without MGO at concentrations of 5, 10 and 15 mM for indicated time period. M - molecular weight marker

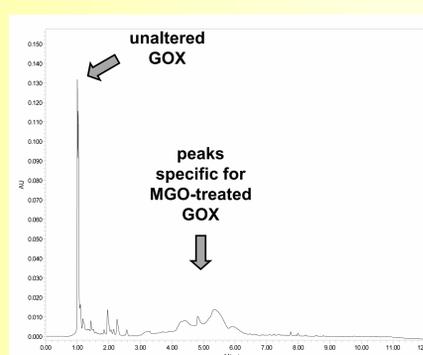
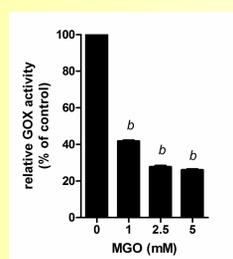


Fig.5 In vitro cross-linking of GOX by MGO - UHPLC analysis. Ten μl of GOX (1 mg/ml) after treatment with 15 mM MGO for 48 h was applied on a Waters Acquity UPLC H-Class system equipped with BEH300 C4 column. Chromatographic separation was carried out using a gradient elution of acetonitrile (ACN)-water solvent system. AU - absorbance units at 220 nm.



In addition, formation of high molecular weight adducts was accompanied by significant loss of its enzymatic activity (Fig.6).

Fig.6 Effect of MGO on enzymatic activity of GOX. GOX (0.1 mg/ml) was incubated with 1, 2.5 and 5 mM MGO at 37°C for 7 days. Control and treated GOX was mixed with glucose and enzymatic activity was examined upon H₂O₂ development. *b* P<0.001 vs. control (0 mM MGO).

CONCLUSION

Our observation suggests that high levels of MGO in manuka honey are responsible for suppressing H₂O₂ generation. These data highlight the detrimental effect of MGO on proteinaceous components, including those contributing to the antibacterial activity of honey.

However, comprehensive clinical trial study should be performed to assess the advantageousness of various selected honeys (manuka, other floral, honeydew) for specific types of wounds.

